Bohr-ing Practice

For each of the following elements
   a. Draw a Bohr-Rutherford diagram of the ion that it would form
   b. Beside it write the symbol including the charge of this ion (ex Ca$^{2+}$, N$^{3-}$)
   c. State if it is an anion or a cation.

1. oxygen
   a. 
   b. 
   c. 

2. magnesium
   a. 
   b. 
   c. 

3. fluorine
   a. 
   b. 
   c. 

4. phosphorus
   a. 
   b. 
   c. 

5. sodium
   a. 
   b. 
   c.